

EOC CHEMISTRY SAMPLE ITEMS GOAL 2 KEY Printable PDF

eoc chemistry sample items goal 2 key is a crucial phrase for students preparing for their End-of-Course (EOC) Chemistry assessments. Achieving success on these exams requires understanding not only the core concepts but also the types of questions that are typically included, especially those aligned with Goal 2. This article aims to provide a comprehensive overview of the key elements related to EOC Chemistry sample items for Goal 2, offering insights into what students can expect and how to best prepare for these items.

Understanding EOC Chemistry Sample Items for Goal 2

To excel on the EOC Chemistry exam, students need to familiarize themselves with sample items that reflect the types of questions aligned with Goal 2. These sample items serve as a mirror of the actual exam content, providing a valuable opportunity to practice and understand the skills and knowledge required.

What is Goal 2 in EOC Chemistry?

Goal 2 focuses on the structure and properties of matter and how they relate to chemical behavior. It emphasizes understanding atomic structure, periodic table trends, chemical bonding, and the classification of matter. Sample items for Goal 2 often test students' ability to:

- Identify atomic and molecular structures from descriptions or models
- Explain periodic trends such as electronegativity, ionization energy, and atomic radius
- Describe different types of chemical bonds and their properties
- Differentiate between elements, compounds, and mixtures

Understanding these core areas is essential for mastering Goal 2 sample items.

Types of Sample Items in Goal 2

EOC Chemistry sample items for Goal 2 typically include various question formats designed to assess conceptual understanding and application skills.

Multiple-Choice Questions

Most sample items are multiple-choice, requiring students to select the best answer from four options. These questions often test:

- Knowledge of atomic structure and periodic trends
- Application of concepts to real-world scenarios
- Analysis of data or diagrams related to chemical properties

Constructed-Response Items

Some sample items require students to construct brief responses, explanations, or calculations. These assess deeper understanding and the ability to communicate scientific reasoning effectively.

Laboratory-Based Items

While less common in multiple-choice formats, some sample items simulate lab scenarios where students interpret experimental data, analyze results, or predict outcomes based on their understanding of matter's structure and properties.

Key Concepts Covered in Goal 2 Sample Items

Success with Goal 2 sample items hinges on a solid grasp of several fundamental concepts. Below are the key areas students should focus on.

Atomic Structure and the Periodic Table

Understanding the basic structure of an atom—including protons, neutrons, electrons—and how this relates to atomic number and mass is essential. Students should be able to:

- Identify atomic number and mass from atomic models or data
- Describe isotopes and their significance
- Use the periodic table to find element properties

Periodic Trends

Questions often involve analyzing trends across periods and down groups, such as:

- Atomic radius
- Electronegativity

- Ionic and atomic size
- Ionization energy

Students should be able to interpret data tables and graphs illustrating these trends.

Chemical Bonding and Molecular Structure

Sample items may assess understanding of:

- Ionic, covalent, and metallic bonds
- Properties of different bond types
- Lewis structures and molecular geometries
- Polarity and intermolecular forces

Classification of Matter

Students should distinguish between elements, compounds, and mixtures, and understand how matter's structure affects its properties.

Strategies for Mastering Goal 2 Sample Items

Effective preparation involves targeted strategies to familiarize oneself with sample items and the underlying concepts.

Practice with Sample Questions

- Use official practice tests and sample items to simulate exam conditions
- Focus on understanding why each answer choice is correct or incorrect
- Time yourself to improve pacing

Deepen Conceptual Understanding

- Review key vocabulary and definitions
- Draw diagrams and models to visualize atomic structures and bonding
- Use flashcards for periodic trends and properties

Analyze Mistakes and Clarify Misconceptions

- Review explanations for questions you answered incorrectly
- Seek additional resources or tutoring for challenging topics

Resources for EOC Chemistry Goal 2 Preparation

Several resources provide valuable practice and explanations for Goal 2 sample items.

Official Practice Tests and Sample Items

- Available through your state's education department or testing agency
- Include real exam questions and detailed answer explanations

Study Guides and Review Books

- Focus on core concepts related to Goal 2
- Include practice questions modeled after sample items

Online Platforms and Tutorials

- Interactive quizzes and video lessons
- Offer immediate feedback on practice items

Conclusion: Mastering Goal 2 Key for EOC Chemistry Success

Achieving a high score on the EOC Chemistry exam requires familiarity with sample items associated with Goal 2. These items test understanding of atomic structure, periodic trends, chemical bonding, and matter classification—all fundamental to a comprehensive grasp of chemistry. By practicing with sample questions, reviewing core concepts, and utilizing available resources, students can confidently approach the exam and demonstrate their mastery of these essential topics. Remember, success is not just about memorizing facts but about understanding the principles that govern the behavior of matter at the atomic and molecular levels. Preparing effectively for Goal 2 sample items will significantly enhance your overall performance on the EOC Chemistry assessment.

EOC Chemistry Sample Items Goal 2 Key: A Comprehensive Guide to Mastering Key Concepts for Success

Preparing for End-of-Course (EOC) Chemistry exams can be daunting, especially when it comes to

understanding what examiners focus on. One of the most critical components of your preparation involves mastering the EOC Chemistry Sample Items Goal 2 Key. This key provides insight into the essential concepts and skills you need to demonstrate to succeed. In this article, we'll unpack the purpose of Goal 2, analyze typical sample items, and offer strategies to help you confidently approach similar questions on your exam.

Understanding the Purpose of Goal 2 in the EOC Chemistry Framework

Before diving into sample items, it's important to understand what Goal 2 encompasses within the EOC Chemistry standards. Generally, these goals are designed to align with the core skills and knowledge necessary for mastery of chemistry concepts.

Goal 2 primarily focuses on:

- Understanding the structure of atoms and the periodic table.
- Applying stoichiometry and balancing chemical equations.
- Understanding chemical reactions and predicting products.
- Interpreting data from experiments or chemical formulas.
- Applying mathematical skills to solve chemistry problems.

The EOC Chemistry Sample Items Goal 2 Key serves as an answer key and guide, illustrating how students should approach these areas and what evidence of understanding is expected.

Breakdown of Typical Sample Items in Goal 2

Sample items associated with Goal 2 tend to test a student's ability to analyze and interpret chemical data, perform calculations, and demonstrate conceptual understanding. Let's examine common types of questions and how to approach them.

- Atomic Structure and the Periodic Table

Sample Question Example:

Given the atomic number and mass number of an element, determine the number of protons, neutrons, and electrons.

Key Skills Tested:

- Understanding atomic structure.
- Basic calculations involving atomic number, mass number, and isotopes.

Strategy:

- Recall that protons = atomic number.
- Neutrons = mass number - atomic number.
- In a neutral atom, electrons = protons.

Sample item tip: Always verify whether the atom is an ion, which may have a different number of electrons.

- Balancing Chemical Equations

Sample Question Example:

Balance the following chemical equation: $C_2H_4 + O_2 \rightarrow CO_2 + H_2O$.

Key Skills Tested:

- Applying the law of conservation of mass.
- Understanding reactants and products.

Strategy:

- Balance elements step-by-step, starting with the most complex molecule.
- Use coefficients instead of subscripts.
- Check that the number of atoms for each element is the same on both sides.

Sample item tip: Practice balancing equations regularly; it's fundamental to understanding chemical reactions.

- Stoichiometry and Calculations

Sample Question Example:

Calculate the mass of water produced when 5 grams of hydrogen gas reacts with excess oxygen.

Key Skills Tested:

- Mole conversions.
- Using molar masses.
- Applying mole ratios from balanced equations.

Strategy:

- Write the balanced equation: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$.
- Convert grams of hydrogen to moles.
- Use mole ratio to find moles of water.
- Convert moles of water to grams.

Sample item tip: Always double-check your units and ensure your calculations follow logical steps.

- Chemical Reactions and Predicting Products

Sample Question Example:

Predict the products of the reaction between sodium hydroxide and hydrochloric acid.

Key Skills Tested:

- Recognizing types of reactions (acid-base neutralization).
- Knowledge of common reaction products.

Strategy:

- Recall that acid-base reactions produce water and a salt.
- Write the neutralization: $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$.

Sample item tip: Familiarize yourself with common acids, bases, and salts to quickly identify products.

- Data Interpretation and Experimental Analysis

Sample Question Example:

An experiment measures the pH of a solution over time. How does the pH change if an acid is added?

Key Skills Tested:

- Understanding pH scale.
- Interpreting experimental data.
- Recognizing acid-base behavior.

Strategy:

- Recall that adding acid decreases pH (more acidic).
- Use data to support your reasoning.

Sample item tip: Practice reading graphs and interpreting data to be comfortable with experimental scenarios.

Strategies for Mastering Goal 2 Sample Items

Achieving mastery in Goal 2-related questions involves more than rote memorization. Here are effective strategies:

- Deepen Conceptual Understanding
- Don't just memorize formulas; understand why formulas work.
- Visualize atomic structures and reaction mechanisms.
- Practice with Past EOC Sample Items
- Use released tests and practice questions to familiarize yourself with question formats.
- Review the Goal 2 Key to understand the rationale behind correct answers.

- Develop Strong Mathematical Skills
- Practice conversions between grams, moles, and particles.
- Master balancing equations and stoichiometry calculations.
- Use Visual Aids and Models
- Draw atomic models, reaction diagrams, and periodic table charts.
- Use online simulations to visualize chemical reactions.
- Focus on Data Analysis
- Practice interpreting tables, graphs, and experimental data.
- Develop critical thinking skills to analyze trends and anomalies.

Common Pitfalls and How to Avoid Them

Even experienced students can fall into traps when tackling Goal 2 items. Be aware of these common pitfalls:

- Ignoring units: Always include units in calculations to prevent errors.
- Misreading questions: Carefully read each question to understand what is being asked.
- Overlooking assumptions: For example, assuming an atom is neutral unless stated otherwise.
- Neglecting to check work: Always review calculations and reasoning.

Final Tips for Success

- Study consistently rather than cramming.
- Create a cheat sheet with key formulas, concepts, and common reaction types.
- Join study groups to discuss and clarify challenging topics.
- Seek help early if concepts aren't clear?don't wait until the last minute.
- Simulate exam conditions by timing yourself on practice questions.

Conclusion

Mastering the EOC Chemistry Sample Items Goal 2 Key is an essential step toward exam success. By understanding the types of questions typically asked, practicing strategic approaches, and reviewing core concepts, students can confidently navigate the complexities of chemistry. Remember, thorough preparation, consistent practice, and a clear understanding of foundational principles will empower you to excel in your EOC Chemistry exam and beyond.

Question What is the main focus of Goal 2 in the EOC Chemistry Sample Items? Goal 2 emphasizes understanding the structure and properties of atoms and molecules, including atomic theory, electron configurations, and chemical bonding. How can students effectively prepare for Goal 2 questions in the EOC Chemistry exam? Students should review atomic models, practice interpreting periodic table trends, and solve sample problems related to chemical bonding and molecular structures. What types of questions are commonly found in Goal 2 of the EOC Chemistry sample items? Questions often involve identifying atomic structures, predicting bond types, drawing Lewis structures, and explaining how atomic properties influence chemical behavior. Why is understanding electron configurations important for Goal 2 in EOC Chemistry? Understanding electron configurations helps students explain the arrangement of electrons in atoms, which is essential for predicting chemical reactivity, bonding patterns, and physical properties. Where can I find reliable resources for practicing Goal 2 key concepts in EOC Chemistry? Resources include official state assessment guides, practice tests, chemistry textbooks, and educational websites like Khan Academy and the College Board's AP Chemistry materials.

Related keywords: EOC Chemistry, Sample Items, Goal 2, Key, Chemistry Assessment, Standard 2, Chemistry Practice Questions, Science Standards, Chemistry Exam Prep, Sample Test Items

Ncdpi Released Test Items 2013

Understanding the Significance of NCDPI Released Test Items 2013

ncdpi released test items 2013 marked a pivotal moment in North Carolina's educational assessment landscape. These released test items provided educators, students, and parents with valuable insights into the types of questions and standards assessed during that academic year. The North Carolina Department of Public Instruction (NCDPI) periodically releases test items to promote transparency, aid in test preparation, and help educators align their instruction with state standards. In 2013, the release included a comprehensive set of items across various subjects and grade levels, offering an in-depth look at the assessment expectations for students at that time.

Understanding the context behind these released items is essential for educators seeking to improve instructional strategies, students preparing for exams, and parents supporting their

children's academic journey. This article explores the details of the 2013 test items, how they can be used effectively, and the broader implications for education in North Carolina.

Overview of NCDPI Test Item Release 2013

Purpose of Releasing Test Items

The NCDPI releases test items for multiple reasons:

- To provide transparency about the content and format of assessments.
- To assist teachers in aligning their lesson plans with state standards.
- To serve as practice resources for students.
- To help stakeholders understand the skill levels expected for various grades and subjects.

In 2013, the released items covered several key areas, including mathematics, English language arts (ELA), science, and social studies, reflecting the state's commitment to comprehensive student assessment.

Subjects Covered and Grade Levels Included

The 2013 release included items from:

- Mathematics (Grades 3-8 and high school)
- English Language Arts (Grades 3-8 and high school)
- Science (Grades 5, 8, and high school)
- Social Studies (Grades 5, 8, and high school)

This wide coverage aimed to give a holistic view of student competencies across core disciplines.

Key Features of the 2013 Test Items

Question Formats and Types

The released test items showcased various question formats designed to assess different cognitive skills:

- Multiple-choice questions
- Constructed-response questions

- Performance-based tasks
- Technology-enhanced items

These formats encouraged critical thinking and application of knowledge, aligning with Next Generation Science Standards and Common Core State Standards.

Content Focus and Standards Alignment

Each test item was aligned with specific North Carolina Standard Course of Study (NCSCOS) objectives. For example:

- Mathematics: Focused on operations, algebra, geometry, data analysis.
- English Language Arts: Emphasized reading comprehension, writing, language conventions.
- Science: Covered scientific inquiry, earth sciences, life sciences.
- Social Studies: Addressed civics, geography, history, economics.

This alignment ensured that the test items accurately reflected the learning goals for each grade level.

How to Use the 2013 Test Items for Effective Preparation

For Educators

Teachers can leverage the released test items to:

- Design Practice Assessments: Create quizzes or practice tests mirroring the format and difficulty.
- Identify Instructional Gaps: Use item analysis to determine which standards need reinforcement.
- Enhance Test-Taking Skills: Familiarize students with question types and testing strategies.
- Align Curriculum: Ensure instructional activities align with tested standards.

For Students

Students preparing for upcoming assessments can:

- Review released items to understand question structure.
- Practice similar questions to build confidence.
- Focus on areas where they show weaknesses based on item analysis.

- Develop test-taking strategies tailored to question formats.

For Parents

Supporting children with their test preparation involves:

- Reviewing sample questions together.
- Encouraging a balanced study routine.
- Understanding the skills assessed to better support learning at home.

Accessing and Utilizing the 2013 Test Items

Where to Find the Released Test Items

The NCDPI website archives released test items, including those from 2013. Educators and parents can access these resources through:

- The NCDPI official assessment resources page.
- State-approved testing portals.
- Educational resource repositories.

Additionally, many educational publishers and tutoring centers incorporate these items into their practice materials.

Using Digital Platforms for Practice

Many online platforms host practice tests modeled after the released items, offering interactive experiences that mimic the testing environment. These platforms often include:

- Immediate feedback
- Progress tracking
- Customizable practice sessions

Utilizing such tools can enhance student readiness and familiarity with test formats.

Impacts of the 2013 Test Items Release on North Carolina Education

Improved Alignment of Instruction and Assessment

The release of test items led to more precise curriculum planning, ensuring assessments accurately reflected classroom instruction and standards.

Enhanced Teacher Professional Development

Teachers used these resources for targeted professional development, focusing on areas like question analysis and standard mastery.

Increased Student Preparedness

Students gained better understanding of assessment expectations, reducing test anxiety and improving performance.

Evolution of Test Item Releases Beyond 2013

While 2013 set a precedent, subsequent years saw:

- Increased use of technology-enhanced items.
- Broader online access to practice resources.
- Alignment with updated educational standards.
- Greater emphasis on critical thinking and problem-solving skills.

Understanding the progression from 2013 helps educators and stakeholders appreciate ongoing efforts to improve assessment practices.

Conclusion: The Legacy and Future of NCDPI Released Test Items

The **ncdpi released test items 2013** continue to serve as valuable educational tools. They provide transparency, help bridge the gap between classroom instruction and assessment, and support a culture of continuous improvement in North Carolina's education system. As assessment techniques evolve, future releases will likely incorporate more interactive and adaptive elements, but the foundational role of these released items in shaping effective teaching and learning remains vital.

For educators, students, and parents, engaging with these resources fosters a deeper understanding of academic expectations and enhances overall student achievement. Staying informed about past and current test item releases ensures that all stakeholders are better prepared to meet the evolving demands of education assessment.

Remember: Accessing the official NCDPI resources regularly and integrating released test items

into your study or teaching routines can significantly impact academic success. Stay proactive, utilize available practice tools, and foster a supportive learning environment to maximize the benefits of these valuable assessment resources.

NCDPI Released Test Items 2013: An In-Depth Analysis of Educational Assessment Materials

In the landscape of educational assessment, the North Carolina Department of Public Instruction (NCDPI) has played a pivotal role in shaping accountability and instructional strategies through the release of test items. The 2013 release of NCDPI test items marked a significant milestone, providing educators, policymakers, and stakeholders with valuable insights into the state's assessment standards, curriculum alignment, and student performance expectations. This comprehensive review aims to dissect the significance of these released test items, their structure, content, and implications for educational practice.

Understanding the Purpose of NCDPI Released Test Items

Facilitating Transparency and Alignment

The primary goal of releasing test items from NCDPI in 2013 was to promote transparency in the assessment process. By providing sample questions and test formats, the department aimed to:

- Help educators understand the types of questions students would encounter.
- Ensure assessments accurately measure the curriculum standards.
- Improve instructional planning by aligning teaching strategies with assessment expectations.

Transparency also served to build trust among educators, parents, and students by clarifying the assessment criteria and expectations.

Supporting Test Preparation and Professional Development

Released test items serve as essential tools for:

- **Test Preparation:** Allowing students to familiarize themselves with question formats and content scope.
- **Teacher Training:** Equipping educators with resources to develop targeted instruction and practice assessments.
- **Curriculum Review:** Assisting curriculum developers in ensuring content aligns with tested standards.

In 2013, these resources became especially crucial as North Carolina transitioned to new standardized testing frameworks, including the Common Core State Standards (CCSS) in some subject areas.

Structure and Content of the 2013 Test Items

Subject Areas and Grade Levels

The released test items covered a broad spectrum of subjects and grade levels, including:

- Mathematics (Grades 3-8, high school end-of-course assessments)
- English Language Arts (ELA) (Grades 3-8, high school)
- Science (Grades 5 and 8, high school courses)
- Social Studies (Grades 4 and 7)

This comprehensive coverage aimed to ensure consistency and alignment across the entire K-12 spectrum.

Question Types and Formats

The 2013 test items featured a variety of question types designed to assess different cognitive skills:

- Multiple-choice questions: The most common format, assessing recall, comprehension, and application.
- Constructed-response questions: Requiring students to generate written responses, demonstrating deeper understanding.
- Performance tasks: More complex items simulating real-world scenarios, especially in science and social studies.
- Technology-enhanced items: Incorporating interactive elements, such as drag-and-drop or simulations, reflecting the increasing integration of technology in assessments.

The diversity in question types aimed to evaluate not only factual knowledge but also critical thinking, problem-solving, and analytical skills.

Content Standards and Cognitive Levels

Each test item was designed to align with specific content standards and cognitive levels outlined by the assessment frameworks. The cognitive levels ranged from:

- Recall and Reproduction: Basic knowledge and memorization.
- Skills and Concepts: Understanding and applying learned concepts.
- Strategic Thinking and Reasoning: Analyzing, evaluating, and synthesizing information.

The 2013 test items emphasized higher-order thinking skills, reflecting a shift toward more rigorous assessment practices aligned with the Common Core and other standards.

Analysis of the 2013 Test Items: Strengths and Challenges

Strengths of the 2013 Released Test Items

- Enhanced Transparency: Providing sample items allowed educators and students to better prepare for actual testing environments.
- Curriculum Alignment: The items closely reflected the standards, ensuring assessments accurately measured intended learning outcomes.
- Instructional Utility: Teachers could use the released items to develop practice tests, formative assessments, and targeted instruction.
- Focus on Critical Thinking: Incorporation of higher-order questions encouraged deeper understanding and critical analysis.

Challenges and Limitations

- Limited Scope: While the released items offered valuable insights, they could not encompass the full breadth of the assessment content, potentially leading to gaps in preparation.
- Overemphasis on Test-Taking Skills: Students might focus on practicing specific question types rather than broad conceptual understanding.
- Accessibility and Equity: Not all students had equal access to practice materials or resources, which could impact performance disparities.
- Evolution of Standards: Since 2013, assessment frameworks have continued to evolve, potentially rendering some test items outdated or less reflective of current standards.

Implications for Educators and Policymakers

For Educators

- Utilize Released Items for Instructional Planning: Analyzing test items helps identify key concepts and skills emphasized in assessments.
- Design Practice Assessments: Teachers can develop classroom tests modeled after released

items to prepare students effectively.

- Focus on Higher-Order Skills: Emphasizing analytical and reasoning questions in instruction aligns with the assessment's focus on critical thinking.

For Policymakers and Curriculum Developers

- Ensure Ongoing Transparency: Regular release of test items fosters trust and continuous improvement.
- Align Curriculum with Assessment Standards: Regular review of test items can guide curriculum adjustments to meet assessment expectations.
- Address Equity Concerns: Provide equitable access to practice materials and resources to support all student populations.

Future Directions and Recommendations

- Update and Expand Released Items: As standards evolve, so should the released test items to reflect current assessment goals.
- Incorporate Technology-Enhanced Items: Embrace digital formats to mirror actual testing environments more accurately.
- Provide Detailed Item Analyses: Offering explanations and rationales for each item can deepen understanding and instructional relevance.
- Facilitate Professional Development: Train educators on effectively integrating released test items into their teaching practices.

Conclusion: The Legacy and Continued Relevance of 2013 Test Items

The NCDPI released test items from 2013 serve as a valuable resource for multiple stakeholders in North Carolina's education system. They exemplify efforts toward transparency, standards alignment, and instructional support. While challenges remain, such as keeping materials up-to-date and ensuring equitable access, the release of these test items has contributed significantly to fostering a culture of informed assessment practices.

Looking ahead, continuous efforts to update and expand upon these resources will be vital in maintaining high standards of educational quality and equity. As assessment practices evolve, so too must the tools and materials educators rely upon, ensuring that student learning remains at the heart of educational improvement initiatives.

References:

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Note: The above references are illustrative; actual sources should be consulted for precise information.

Question What is the significance of the NCDPI released test items from 2013? The NCDPI released test items from 2013 serve as valuable resources for educators and students to understand the types of questions, standards, and content that were assessed during that year, aiding in effective preparation and curriculum alignment. Where can I access the 2013 NCDPI released test items? The 2013 NCDPI released test items can typically be accessed through the North Carolina Department of Public Instruction's official website or their archived resources section dedicated to past assessments. How can educators use the 2013 NCDPI test items to improve instruction? Educators can analyze the test items to identify key concepts, question formats, and skills emphasized that year, allowing them to tailor instruction to better prepare students for similar assessments. Are the 2013 NCDPI test items aligned with current standards? While some foundational concepts remain consistent, it is important to cross-reference the 2013 test items with current standards, as educational standards may have evolved since then. What subjects are covered in the 2013 NCDPI released test items? The released test items from 2013 typically cover core subjects such as Mathematics, English Language Arts, Science, and Social Studies, reflecting the assessments administered that year. Can I use the 2013 test items for practice tests? Yes, educators and students can use these test items as practice tests to familiarize themselves with question formats and to assess understanding of the tested content from 2013. Are there any known changes in testing policies since 2013 related to these test items? Yes, testing policies and standards may have changed since 2013, so it's important to review current guidelines and standards to ensure relevancy when using these test items for preparation. How do the 2013 test items reflect the educational priorities of that year? The 2013 test items highlight the focus areas and skills emphasized by the North Carolina education system at that time, providing insight into the curriculum priorities and assessment strategies used. Is there a way to compare 2013 test items with recent assessments to track progress? Yes, educators and researchers can analyze and compare test items from 2013 with recent assessments to identify trends, improvements, and areas needing reinforcement in student learning.

Related keywords: NCDPI, test items, 2013, North Carolina Department of Public Instruction, standardized testing, test questions, assessment items, exam blueprint, state assessments, curriculum standards

Read the full article online: [Ncdpi Released Test Items 2013](#)